

Fig. 1. Molecular structure and atom numbering. *ORTEP* (Johnson, 1965) was used for this drawing and for Fig. 3.

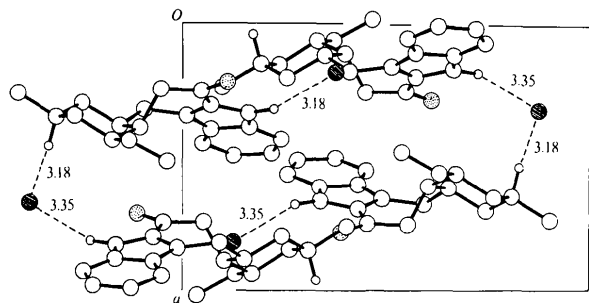


Fig. 3. Crystal structure viewed along *b*.

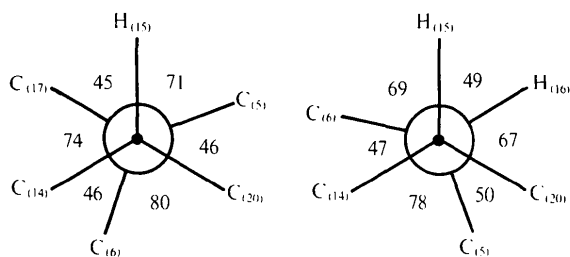


Fig. 2. Newman projections along C(15)–C(16). (a) Ervatamine. (b) This work.

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*Acta Cryst.* (1977). **B33**, 135–137

## Trimethyltin Methylsulphinate

BY GEORGE M. SHELDRIK AND ROBIN TAYLOR

*University Chemical Laboratory, Lensfield Road, Cambridge CB2 1EW, England*

(Received 23 June 1976; accepted 3 July 1976)

**Abstract.**  $C_4H_{12}O_2SSn$ , orthorhombic,  $C22_2$ ;  $a = 17.569$  (20),  $b = 12.293$  (20),  $c = 8.240$  (12) Å,  $U = 1778$  Å<sup>3</sup>,  $Z = 8$ ,  $D_x = 1.81$  g cm<sup>-3</sup>,  $\mu(Mo K\alpha) = 28.18$  cm<sup>-1</sup>. The structure was solved by the heavy-atom method and refined to an  $R$  of 0.043 for 719 unique two-circle diffractometer data. Planar trimethyltin groups are linked by disordered –O–S(Me)–O– units to form infinite helical chains along *c*, with trigonal bipyramidal geometry about Sn.

**Introduction.** Several tetraorganotin derivatives undergo insertion of  $SO_2$  into the Sn–C bond to give stable, crystalline mono-inserted products which have

been shown by IR and NMR studies to be aggregated in solution (Fong & Kitching, 1970). In view of the facility with which Sn increases its coordination number beyond four when a suitable bridging ligand is available, it seemed likely that these insertion compounds are polymeric in the solid state, and this has been confirmed by the structure of  $Me_3SnOS(O)CH_2CCH$  (Ginderow & Huber, 1973) and by the determination reported here.

The sample was prepared by dissolving  $Me_4Sn$  in dry liquid  $SO_2$  under vacuum. Evaporation of excess  $SO_2$  left a white solid which was recrystallized from  $CCl_4$  to afford a few plate-like crystals which were mounted in

Lindemann glass capillary tubes. Intensities were determined with an automated Stoe two-circle diffractometer, Mo  $K\alpha$  radiation, graphite monochromator, and a crystal  $0.04 \times 0.25 \times 0.11$  mm (layers  $h0-12l$ ). 1377 reflexions were measured in an approximately constant count mode; Lp and absorption corrections were applied, and equivalent reflexions averaged to give 719 unique observed data.  $a$  and  $c$  were obtained by a least-squares fit to the  $\omega$  values of 141  $h0l$  reflexions, and  $b$  was obtained from diffractometer  $\mu$  measurements.

The atoms were located from Patterson and difference syntheses; the MeS moiety was found to be disordered, and was refined with the site occupation factors constrained so that  $k_a + k_b = 1$ , where  $k_a$  is the value for both S(1a) and C(5a), and  $k_b$  for S(1b) and C(5b).  $k_a$  refined to 0.55 (2); common temperature factors were employed for S(1a) and S(1b), and for C(5a) and C(5b). Complex neutral-atom scattering factors were employed; the weighting scheme was  $w = \Sigma[\sigma^2(N) + 0.001N^2]^{-1}/(30.13 + |F_o| + 0.00551F_o^2)$ , where  $N$  is the net peak count, and the summation is over equivalent reflexions. 58 parameters were varied independently in the full-matrix least-squares refinement, including interlayer scale factors, anisotropic temperature factors for Sn and isotropic temperature factors for the light atoms. To avoid correlation between the interlayer scale factors and the  $U_{22}$  components of the Sn temperature factors, the latter were constrained to be equal

to the mean of  $U_{11}$  and  $U_{33}$ . The refinement converged to  $R' = \Sigma w^{1/2} \Delta / \Sigma w^{1/2} |F_o| = 0.046$  and  $R = 0.043$ ; refinement in the opposite enantiomorph produced no

Table 3. Bond lengths ( $\text{\AA}$ )

Sn(1)—O(1)	2.209 (31)	Sn(2)—O(2)	2.272 (19)
Sn(1)—C(1)	2.137 (30)	Sn(2)—C(3)	2.104 (37)
Sn(1)—C(2)	2.113 (23)	Sn(2)—C(4)	2.128 (23)
Disordered group			
S(1a)—O(1)	1.467 (29)	S(1b)—O(1)	1.354 (30)
S(1a)—O(2)	1.468 (21)	S(1b)—O(2)	1.384 (20)
S(1a)—C(5a)	1.565 (53)	S(1b)—C(5b)	1.561 (59)

Table 4. Bond angles ( $^\circ$ )

O(1)—Sn(1)—O(1')	179.1 (12)	O(2)—Sn(2)—O(2')	172.9 (8)
C(1)—Sn(1)—O(1)	89.6 (6)	C(3)—Sn(2)—O(2)	86.4 (6)
C(2)—Sn(1)—O(1)	88.0 (9)	C(4)—Sn(2)—O(2)	91.5 (8)
C(2)—Sn(1)—C(1)	120.4 (6)	C(4)—Sn(2)—C(3)	119.6 (7)
C(2)—Sn(1)—C(2')	119.2 (11)	C(4)—Sn(2)—C(4')	120.7 (12)
Disordered group			
O(2)—S(1a)—O(1)	108.3 (14)	O(2)—S(1b)—O(1)	120.7 (15)
C(5a)—S(1a)—O(1)	110.1 (24)	C(5b)—S(1b)—O(1)	126.1 (26)
C(5a)—S(1a)—O(2)	126.3 (25)	C(5b)—S(1b)—O(2)	110.6 (25)
S(1a)—O(1)—Sn(1)	140.4 (15)	S(1a)—O(2)—Sn(2)	130.4 (12)
S(1b)—O(1)—Sn(1)	169.2 (17)	S(1b)—O(2)—Sn(2)	146.2 (13)

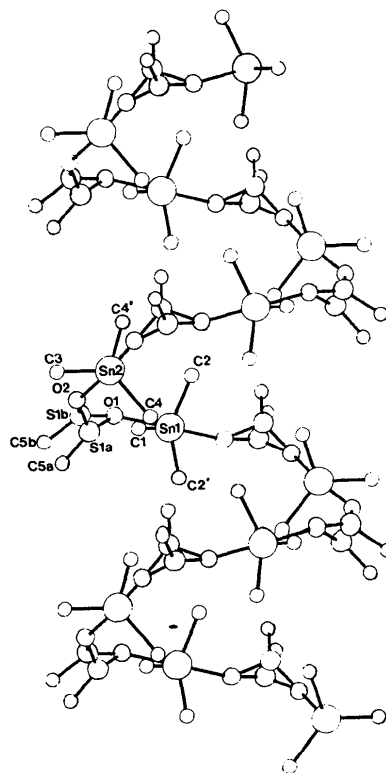
Table 1. Atom coordinates ( $\times 10^4$ ) and isotropic temperature factors ( $\text{\AA}^2 \times 10^3$ )

	$x$	$y$	$z$	$U$
Sn(1)	-1750 (1)	0	0	
Sn(2)	0	-3311 (1)	2500	
O(1)	-1760 (10)	-1757 (23)	576 (27)	133 (7)
O(2)	-1158 (9)	-3426 (15)	1283 (19)	106 (5)
C(1)	-2967 (15)	0	0	82 (8)
C(2)	-1142 (11)	253 (16)	2179 (25)	82 (6)
C(3)	0	-5024 (28)	2500	56 (7)
C(4)	460 (11)	-2454 (17)	482 (24)	83 (6)
Disordered atoms				
S(1a)	-1481 (7)	-2797 (7)	-66 (15)	57 (2)
S(1b)	-1763 (8)	-2859 (9)	621 (18)	57 (2)
C(5a)	-2038 (28)	-3236 (49)	-1397 (63)	109 (12)
C(5b)	-2261 (33)	-3624 (50)	-457 (71)	109 (12)

Table 2. Anisotropic temperature factors ( $\text{\AA}^2 \times 10^3$ )

The temperature factor exponent takes the form:  
 $-2\pi^2(U_{11}h^2a^{*2} + \dots + 2U_{12}hka^*b^*)$ .

	$U_{11}$	$U_{22}$	$U_{33}$	$U_{23}$	$U_{13}$	$U_{12}$
Sn(1)	51 (1)	63 (1)	73 (1)	13 (1)	0	0
Sn(2)	63 (1)	59 (1)	55 (1)	0	-5 (1)	0

Fig. 1. The structure of trimethyltin methylsulphinate, illustrating the SMe-group disorder and a helix parallel to  $c$ .

significant changes in the  $R$  indices. Final atomic coordinates and thermal parameters are given in Tables 1 and 2, and the resulting bond lengths and angles in Tables 3 and 4.\*

**Discussion.** Trimethyltin moieties are linked by disordered  $-\text{O}-\text{S}(\text{Me})-\text{O}-$  groups to form infinite helical chains along  $c$ , with approximately trigonal bipyramidal geometry at Sn (Fig. 1). There are two crystallographically independent  $\text{Me}_3\text{Sn}$  groups, each arranged with one of the Sn—C bonds along a crystallographic twofold axis so that the group is exactly planar and the Sn atom is coordinated symmetrically by the O atoms to which it is bonded. Symmetrical bridging is also found in  $\text{Me}_3\text{SnOS}(\text{O})\text{CH}_2\text{CCH}$  [Ginderow & Huber (1973); Sn—O 2.306 (10) and 2.296 (10) Å] where, however, the chairs adopt a zigzag rather than helical conformation. The Sn—O distances in the methylsulphinate are similar to those in trimethyltin methoxide [2.20 (2) and 2.26 (2)

Å; Domingos & Sheldrick (1974)], but shorter than the values quoted above for the  $-\text{OS}(\text{O})\text{CH}_2\text{CCH}$  derivative; this, and the relatively short Sn—C bonds, may arise from librational effects. Subsequent inspection of the X-ray photographs revealed the presence of weak diffuse low-order reflexions corresponding to a  $P$  rather than a  $C$  lattice, indicating some short-range order in the conformation adopted by the disordered SME groups.

We are grateful to the Science Research Council for providing the diffractometer and for financial support to RT. The calculations were performed on the Cambridge University IBM 370/165 computer with programs written by GMS. Fig. 1 was drawn with *PLUTO* written by Dr W. D. S. Motherwell.

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\* A list of structure factors has been deposited with the British Library Lending Division as Supplementary Publication No. SUP 31977 (6 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 13 White Friars, Chester CH1 1 NZ, England.

*Acta Cryst.* (1977). B33, 137–139

### Methyl(triphenylphosphine)gold

BY PAUL D. GAVENS, JOHN J. GUY, MARTIN J. MAYS AND GEORGE M. SHELDRIK  
*University Chemical Laboratory, Lensfield Road, Cambridge CB2 1EW, England*

(Received 23 June 1976; accepted 3 July 1976)

**Abstract.**  $\text{C}_{19}\text{H}_{18}\text{AuP}$ , monoclinic,  $P2_1/n$ ,  $a = 17.29$  (1),  $b = 11.27$  (1),  $c = 8.92$  (1) Å,  $\beta = 104.5$  (1)°,  $U = 1682$  Å<sup>3</sup>,  $Z = 4$ ,  $D_x = 1.88$  g cm<sup>-3</sup>. There is essentially linear coordination of gold [P—Au—C = 179.1 (8)°] with Au—C 2.124 (28) and Au—P 2.279 (8) Å. The structure was refined to an  $R$  of 0.092 for 1867 unique diffractometer data.

**Introduction.** Previous crystallographic studies of Au<sup>I</sup> complexes containing Au—C bonds (Bellon, Manassero & Sansoni, 1969; Corfield & Shearer, 1967; Baker & Pauling, 1972) have involved  $sp$  or  $sp^2$  hybridized C, in which  $\pi$ -bonding between Au<sup>I</sup> and C is possible. The Au<sup>I</sup>—C( $sp^3$ ) distance in the title compound provides a suitable reference bond length in which  $\pi$ -bonding is absent.

Colourless crystals were obtained by recrystalliza-

tion from benzene/light petroleum of a sample obtained by standard methods (Tamaki & Kochi, 1973), and mounted in Lindemann glass capillary tubes. Intensities were measured on a Stoe automated two-circle diffractometer, with Mo  $K\alpha$  radiation ( $\lambda = 0.71069$  Å) and a graphite crystal monochromator. Cell dimensions were obtained by a least-squares fit to 652  $hk0$  and 444  $h0l$  diffractometer zero-layer  $\omega$ -angle measurements. The crystals possessed large mosaic spreads and the data were clearly of mediocre quality; only the  $00l$  data (layers  $hk0-10$ ) were used for structure refinement.  $L_p$  and empirical absorption corrections were applied. 2580 reflexions were measured; averaging equivalent reflexions gave 1867 reflexions with  $I > 3\sigma(I)$  based on counting statistics.

The atoms were located from Patterson and difference syntheses; the structure was refined by blocked-